

## Periodic chart of elements



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|--|--------------------------|-----------------------------------------------|
|  | Metals                   | <input type="button" value="Select element"/> |
|  | Semi-conductors          | <input type="button" value="Select element"/> |
|  | Non-metals               | <input type="button" value="Select element"/> |
|  | Inert gasses             | <input type="button" value="Select element"/> |
|  | Lanthanides en actinides | <input type="button" value="Select element"/> |



Each chemical element contains a link to a page that explains its [chemical properties](#), [health effects](#), [environmental effects](#), application data, an image and also information of the history/inventor of each element.

I	II	Choose elements by <a href="#">name</a> , by <a href="#">atomic number</a> , by <a href="#">symbol</a> , by <a href="#">mass</a>												III	IV	V	VI	VII	VIII				
1	<a href="#">H<sub>1</sub></a>	<a href="#">Li<sub>3</sub></a>	<a href="#">Be<sub>4</sub></a>	<a href="#">Na<sub>11</sub></a>	<a href="#">Mg<sub>12</sub></a>	<a href="#">K<sub>19</sub></a>	<a href="#">Ca<sub>20</sub></a>	<a href="#">Sc<sub>21</sub></a>	<a href="#">Ti<sub>22</sub></a>	<a href="#">V<sub>23</sub></a>	<a href="#">Cr<sub>24</sub></a>	<a href="#">Mn<sub>25</sub></a>	<a href="#">Fe<sub>26</sub></a>	<a href="#">Co<sub>27</sub></a>	<a href="#">Ni<sub>28</sub></a>	<a href="#">Cu<sub>29</sub></a>	<a href="#">Zn<sub>30</sub></a>	<a href="#">Ga<sub>31</sub></a>	<a href="#">Ge<sub>32</sub></a>	<a href="#">As<sub>33</sub></a>	<a href="#">Se<sub>34</sub></a>	<a href="#">Br<sub>35</sub></a>	<a href="#">Kr<sub>36</sub></a>
2	<a href="#">B<sub>5</sub></a>	<a href="#">C<sub>6</sub></a>	<a href="#">N<sub>7</sub></a>	<a href="#">O<sub>8</sub></a>	<a href="#">F<sub>9</sub></a>	<a href="#">Ne<sub>10</sub></a>																	
3	<a href="#">Al<sub>13</sub></a>	<a href="#">Si<sub>14</sub></a>	<a href="#">P<sub>15</sub></a>	<a href="#">S<sub>16</sub></a>	<a href="#">Cl<sub>17</sub></a>	<a href="#">Ar<sub>18</sub></a>																	
4	<a href="#">Rb<sub>37</sub></a>	<a href="#">Sr<sub>38</sub></a>	<a href="#">Y<sub>39</sub></a>	<a href="#">Zr<sub>40</sub></a>	<a href="#">Nb<sub>41</sub></a>	<a href="#">Mo<sub>42</sub></a>	<a href="#">Tc<sub>43</sub></a>	<a href="#">Ru<sub>44</sub></a>	<a href="#">Rh<sub>45</sub></a>	<a href="#">Pd<sub>46</sub></a>	<a href="#">Ag<sub>47</sub></a>	<a href="#">Cd<sub>48</sub></a>	<a href="#">In<sub>49</sub></a>	<a href="#">Sn<sub>50</sub></a>	<a href="#">Sb<sub>51</sub></a>	<a href="#">Te<sub>52</sub></a>	<a href="#">I<sub>53</sub></a>	<a href="#">Xe<sub>54</sub></a>					
5	<a href="#">Cs<sub>55</sub></a>	<a href="#">Ba<sub>56</sub></a>	<a href="#">La<sub>57</sub></a>	<a href="#">Hf<sub>72</sub></a>	<a href="#">Ta<sub>73</sub></a>	<a href="#">W<sub>74</sub></a>	<a href="#">Re<sub>75</sub></a>	<a href="#">Os<sub>76</sub></a>	<a href="#">Ir<sub>77</sub></a>	<a href="#">Pt<sub>78</sub></a>	<a href="#">Au<sub>79</sub></a>	<a href="#">Hg<sub>80</sub></a>	<a href="#">Tl<sub>81</sub></a>	<a href="#">Pb<sub>82</sub></a>	<a href="#">Bi<sub>83</sub></a>	<a href="#">Po<sub>84</sub></a>	<a href="#">At<sub>85</sub></a>	<a href="#">Rn<sub>86</sub></a>					
6	<a href="#">Fr<sub>87</sub></a>	<a href="#">Ra<sub>88</sub></a>	<a href="#">Ac<sub>89</sub></a>	<a href="#">Rf<sub>104</sub></a>	<a href="#">Db<sub>105</sub></a>	<a href="#">Sg<sub>106</sub></a>	<a href="#">Bh<sub>107</sub></a>	<a href="#">Hs<sub>108</sub></a>	<a href="#">Mt<sub>109</sub></a>	<a href="#">Ds<sub>110</sub></a>	<a href="#">Uuu<sub>111</sub></a>	<a href="#">Uub<sub>112</sub></a>	<a href="#">Uut<sub>113</sub></a>	<a href="#">Uuq<sub>114</sub></a>	<a href="#">UUp<sub>115</sub></a>	<a href="#">Uuh<sub>116</sub></a>	<a href="#">Uus<sub>117</sub></a>	<a href="#">Uuo<sub>118</sub></a>					
7	<a href="#">Ce<sub>58</sub></a>	<a href="#">Pr<sub>59</sub></a>	<a href="#">Nd<sub>60</sub></a>	<a href="#">Pm<sub>61</sub></a>	<a href="#">Sm<sub>62</sub></a>	<a href="#">Eu<sub>63</sub></a>	<a href="#">Gd<sub>64</sub></a>	<a href="#">Tb<sub>65</sub></a>	<a href="#">Dy<sub>66</sub></a>	<a href="#">Ho<sub>67</sub></a>	<a href="#">Er<sub>68</sub></a>	<a href="#">Tm<sub>69</sub></a>	<a href="#">Yb<sub>70</sub></a>	<a href="#">Lu<sub>71</sub></a>									

<a href="#">Th<sub>90</sub></a>	<a href="#">Pa<sub>91</sub></a>	<a href="#">U<sub>92</sub></a>	<a href="#">Np<sub>93</sub></a>	<a href="#">Pu<sub>94</sub></a>	<a href="#">Am<sub>95</sub></a>	<a href="#">Cm<sub>96</sub></a>	<a href="#">Bk<sub>97</sub></a>	<a href="#"> Cf<sub>98</sub></a>	<a href="#">Es<sub>99</sub></a>	<a href="#">Fm<sub>100</sub></a>	<a href="#">Md<sub>101</sub></a>	<a href="#">No<sub>102</sub></a>	<a href="#">Lr<sub>103</sub></a>					
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An interactive, printable extended version of the Periodic table of chemical elements of Mendeleiev (who invented the periodic table).

(The above picture of the periodic system is interactive -no need to download, just click on an element. For schools and universities please tellchemistry students, teachers and professors to feel free to reference with citation and link for educational purposes)